Hurry-up and Wait! Accelerated Microbiological Shelf-life Studies for Foods.



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A Bit of Background

- Who in the world is Adam Borger?
 - B.S. in Food Science from *THE* Ohio State University.





- M.S. in Food Microbiology from the University of Wisconsin-Madison.
- Microbiologist for Kraft Foods (1998 2009).
- Microbiologist for Rich Products (2009-2013).
- Outreach Program Manager for University of Wisconsin's Food Research Institute (2013 ?).



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Topics

- What's the goal?
- What are you determining?
- What could go correctly.
- Questions to ask before and after your study..

What is the goal?





What is the goal?



- Place product under conditions that accelerate the "reactions" you are monitoring in a food product.
 - Typically, higher temperatures.
- Use this information to determine shelf-life of the food lower temperatures.
- Use these studies or conditions to predict problems after products are in commerce.
- Use these studies to determine if product should be released into commerce.

What are you determining?

- Shelf-life based on:
 - Spoilage
 - Product appearance during temperature abuse.
 - Fat separation; syneresis; other quality defects.
 - Oxidation.
 - Rancidity.
 - Microbiological limits.
 - Pathogen Growth
 - Time to a certain level of microbes in the food.
 - Time to toxin formation.
 - Perhaps time to cell death?

Examples

- Aseptic beverages, puddings, beverages, other items.
 - Somewhat routine to store at elevated temperatures and look for bloated packages.
 - Indicative of contaminants present.
 - Pretty sound temperature abuse and observational conditions.
 - These are typically stored at ambient temperature so what's the correct accelerated temperature?
- Cook-in-bag or hot-filled products.
 - A little more difficult than above.
 - Process cheese chubs, Salad dressings, etc.
 - What might grow at elevated temperatures?
 - Spores? Would this actually happen at lower temperatures?
- Ready-to-eat, open exposed products.
 - Sliced meats, cheeses, cold-packed cheese, yogurt, salads/kits, etc.
 - Probably the most difficult. So many possible contaminants.
- What about raw products?

What could go correctly?

- You will arrive at a dependable set of data that will allow you to <u>easily</u> predict the shelf-life of a food product and <u>the exact moment</u> when the product may suffer microbiological spoilage.
- This dependable and infallible set of data will reduce days, weeks, and months of challenge studies to predict microbiological growth thus saving your company thousands of dollars and earning **YOU** high praise amongst your colleagues.



Questions to consider when performing a study.

- What happens to the product during storage at elevated temperatures?
 - Moisture migration.
 - Mold growth? Bacterial growth?
 - Will other antimicrobial constituents migrate with this moisture?
 - Fat separation.
 - Does this impact behavior of microbes in the product?
 - Different phases?
 - Gums, water binding affected?

Questions to consider when performing a study.

- How can you know which spoilage microbes are of interest?
- Example: Ready-to-Eat sliced meat product
 - What might spoil this product?
 - Lactic acid bacteria
 - Yeasts
 - Molds
 - Gram negative microbes
 - Spore-formers

Questions to consider when performing a study.

- Where are these spoilages microbes coming from?
- Can you be sure that they are all accounted for? Do you need to?
- How do these different microbes behave in your product at elevated temperatures?
- Are these the same microbes that would grow in your product at a lower temperature?
 - Might there be some that grow faster than others?
 - Can you use these to predict shelf-life?





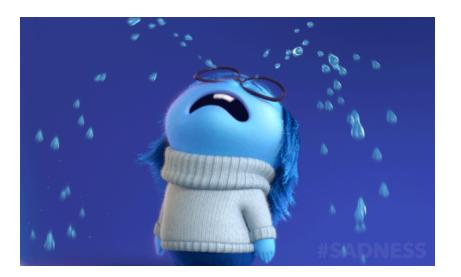
- Birthday party for our 8 year-old (Design of study)
 - We will examine how our scenario impacts the shelf-life (time-to-meltdown) of the product (birthday party).
- Mom (Susan) says "2 hours max."
 - Dad (Adam) says "Oh, 3 hours would be fine."
 - 1.5 times the anticipated shelf-life.



- Daughter's neighborhood friends invited.
 - Known population of organisms. Easily identifiable behavior and known response in environment during normal conditions based on history.
- Daughter's one daycare friend also invited. Mom drops her off and is gone for 3 hours (Adam says "Oh, 3 hours would be fine.")
 - An organism that we don't really know or understand enters the picture.
 - How will this organism react in our environment?



- Birthday party for 8 year-old is underway (Monitoring Results)
- August weather turns very windy and a soaking rain falls; no bouncehouse, everyone in the house – for 3 hours.
 - Accelerated conditions begin to have an effect.
 - The daycare friend (Ava) begins to meltdown.
 - "I HATE juice!" "This is the worst party ever" through tears and sobbing.
 - The party is spiraling out of control at hour 2!



- By hour three Ava is still crying, our daughter is crying, other kids are either indifferent or angry.
 - The experiment may have been a success! We have learned that the accelerated conditions along with an unknown organism have indeed spoiled everything by hour 2.
 - But, what would have happened without the presence of Ava?
 - Do we dare repeat this study?



- At the end of the party/experiment Mom and Dad are drinking...to their success???
 - Susan (Adam's wife) uses this experiment as the standard for setting time limits on birthday parties from here on out – no longer than 1.5 hours as a conservative approach!
 - Adam swears to always listen to Susan....but is this faulty logic?
 - If Ava is not present (and hasn't been since!) do we really need to cut the party so short?

Experimenting with accelerated shelf-life.

- Pick your temperatures.
 - Be sure to include your standard storage temperature for the product.
 - How long will this take?
- Determine your number of trials.
 - This may cause some confusion depending on results.
- Be aware of changes in formulation.
 - Do you have "worst-case" scenario data?
 - Is your new low-salt, clean-label, preservative-free item the "worst-case"?
 - Can you repeat all of these?

What counts as spoiled?



Setting up the experiment – spoilage determination

- What is spoiled?
 - 10⁶ cfu/g?
 - Sight, smell.
 - Is anyone tasting these?
 - Could you accidentally give someone *Clostridium botulinum, C. perfringens of Staphylococcus aureus* poisoning if they did? How would you know?



Can you inoculate your product with isolates from previous incidents?



Q10 – Temperature Coefficient.

- Used for measuring enzymatic reactions.
- The factor by which the rate of spoilage increases when the temperature is raised by 10C.
- Allows for the prediction of a product's shelf life under real-life conditions based on the results of testing conducted at higher temperatures.
- It is unit-less and can be calculated with the equation:
 - Q10 = (R2/R1) (10/(T2-T1))
 - R is the time it takes for a product to spoil and T is the temperature at which the testing is conducted.

Q10 – Hypothetical Example

- Vacuum-packed meal kit is tested at three different temperatures (T).
 - T1 = 4C; T2 = 14C; T3 = 24C
- Days to spoilage (R) are as follows:
 - R1 (4C) = 60 days
 - R2 (14C) = 30 days
 - R3 (24C) = 15 days
- $Q10 = (R2/R1)^{(10/(T2-T1))}$
 - $(30/60)^{(10/(14-4))} = (0.5)^{(10/10)} = 0.5$
 - $(15/30)^{(10/(24-14))} = (0.5)^{(10/10)} = 0.5$
- <u>Dividing</u> the *accelerated shelf-life days to spoilage* by the Q10 value predicts the lower temperature shelf-life:
 - 30 days divided by 0.5 = 60 days at 4C
 - 15 days divided by 0.5 = 30 days. Divide again by 0.5 = 60 days

Q10 – More Complicated Results

- Vacuum-packed meal kit is tested at three different temperatures (T).
 - T1 = 4C; T2 = 14C; T3 = 24C
- Days to spoilage (R) are as follows:
 - R1 (4C) = 60 days
 - R2 (14C) = 25 days
 - R3 (24C) = 3 days
- $Q10 = (R2/R1)^{(10/(T2-T1))}$
 - $(25/60)^{(10/(14-4))} = (0.42)^{(10/10)} = 0.42$
 - $(3/25)^{(10/(24-14))} = (0.12)^{(10/10)} = 0.12$

Q10 – More Complicated Results

• Days to spoilage (R) are as follows:

- R1 (4C) = 60 days• R2 (14C) = 25 days• R3 (24C) = 3 days Q10 = 0.42
- Dividing the accelerated shelf-life value by the Q10 value predicts the lower temperature shelf-life:
 - 25 days divided by 0.42 = 60 days at 4C
 - 3 days divided by 0.12 = 25 days at 14C.
 - Divide again by 0.12 = 208 days at 4C
 - 3 days divided by 0.42 = 7 days at 14C.
 - Divide again by 0.42 = 17 days at 4C

Q10 – More Complicated Results

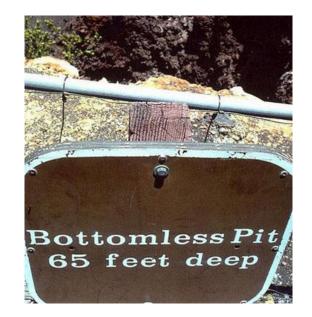
• Days to spoilage (R) are as follows:



- Dividing the accelerated shelf-life value by the Q10 value predicts the lower temperature shelf-life:
 - 3 days divided by 0.22 = 13.6 days at 14C.
 - Divide again by 0.22 = 62 days at 4C

Who is ultimately going to use this information?

- Sales and marketing.
 - How do you explain this to them?
 - What happens if something goes wrong?
- Manufacturing locations.
- Other microbiologists.
- Can you be clear in your message?





Are there other sources of this information?

- Combase and other modeling sites perhaps.
- Industry studies or laboratory studies yes, but do you have access?
- Some dude sticking a product on his desk at the plant and looking at it every day.

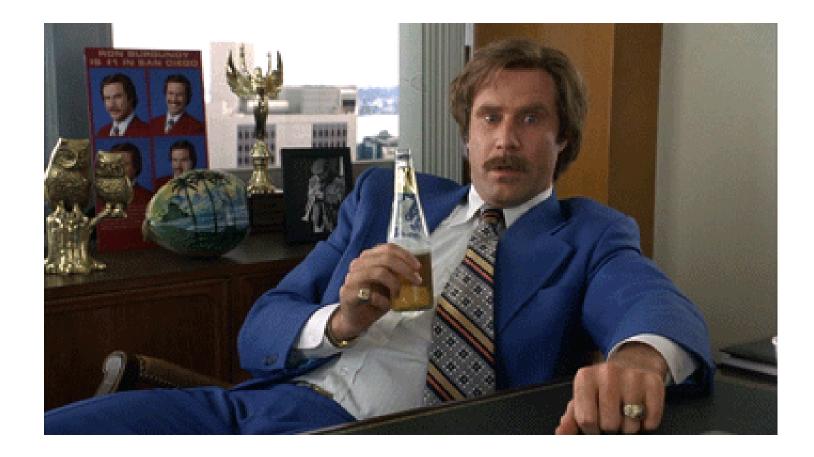


Are there other sources of this information?

- Often VERY specific to your product.
- Perhaps pathogen data can be used from models and other published data.
 - Spoilage is very, very difficult to predict based on all of the factors stated beforehand.

In summary

- Accelerated microbiological shelf-life studies can be performed!
 - Depending on the product, the microbes of interest and a host of other factors.
- Predicting spoilage of products consistently and reliably can be very difficult.
 - Depending again on the host of factors mentioned above.
- Communicating results and possible future trends may be the most difficult aspect of all.
- Don't ever host a child's birthday party for more than 1.5 hours.



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